



Born: Eocene to Holocene era

Education: c.a. 40 MM years of evolution

Positions: Diverse appointments held in CA, OR, and WA

Research interests: Extremely broad, including enzyme-mediated synthesis, reactivity of molten carbon, and CO₂ fixation.

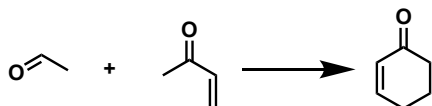
Take-home learnings from this group meeting

1. The purported motivation of cascades is to increase synthetic efficiency, but rarely is this goal achieved.
2. Cascade reactions are generally extremely substrate-dependent, and rarely lead to the discovery of novel broad-scope reactivity.
3. The use of cascade reactions in industry (at bench- or process-scale) is minimal.
4. Cascade reactions can be binned into a small number of reactivity paradigms, and quickly understood on that basis.

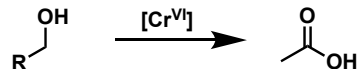
Cascade reaction: A rigorous definition

A cascade (also called tandem, also called domino) reaction must satisfy three criteria:

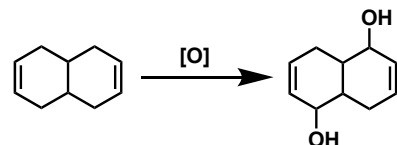
1. Multiple transformations are achieved in a single pot
2. In this single pot, a single set of reaction conditions is maintained
3. Each transformation is dependent on the preceding transformation



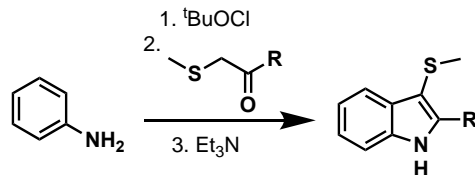
formally, a cascade



formally, a cascade



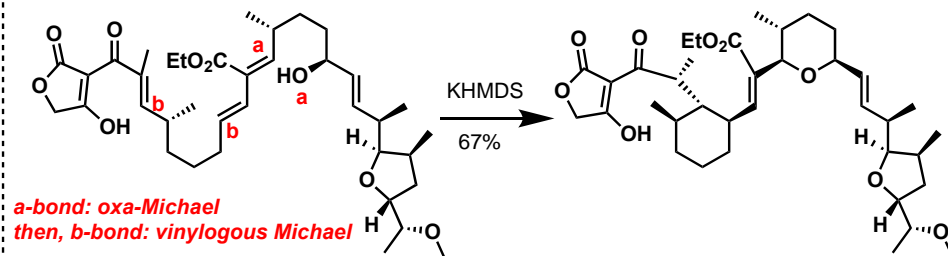
polyfunctionalizations are not cascades



One-pot telescoping is not a cascade

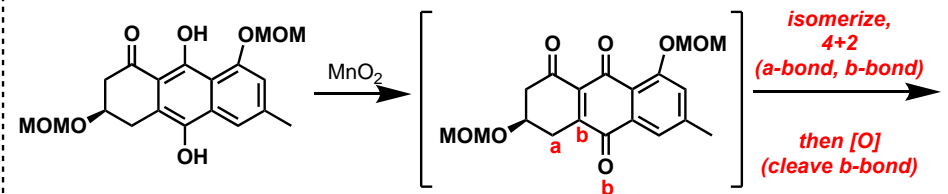
Bin 1: Anionic cascades

Common theme: Michael / Oxa-Michael / Aldol reactivity



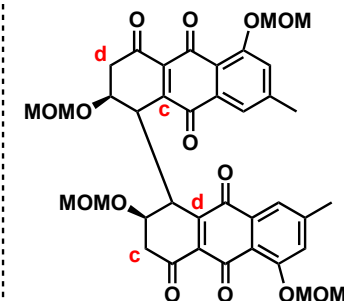
*a-bond: oxa-Michael
then, b-bond: vinylogous Michael*

Tetronasin, Ley et al., *J. Chem Soc., Perkin Trans.*, **1998**, 1, 2259.



*isomerize,
4+2
(a-bond, b-bond)

then [O]
(cleave b-bond)*

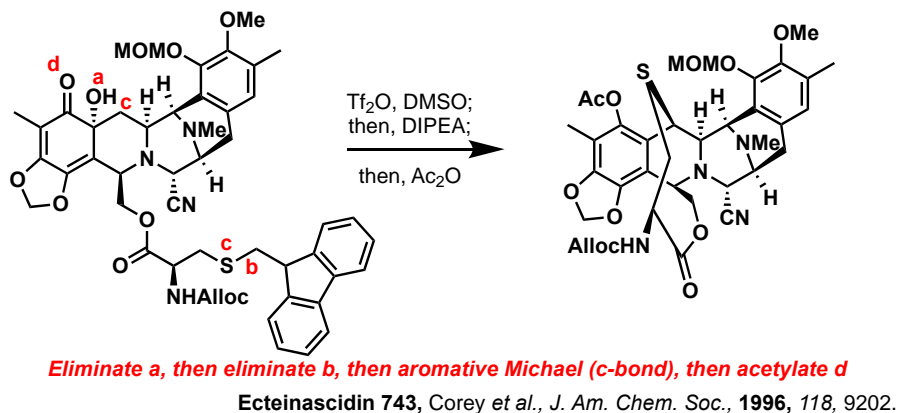
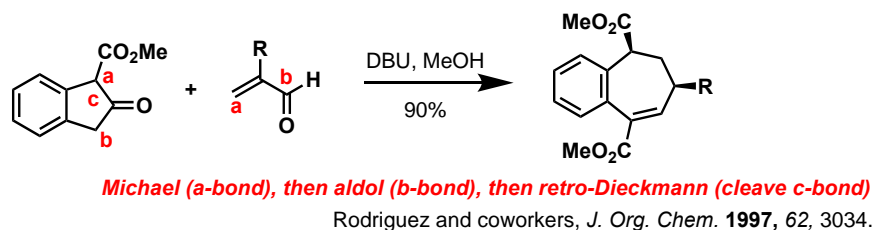
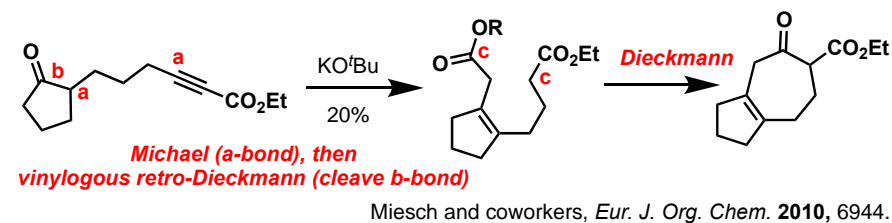
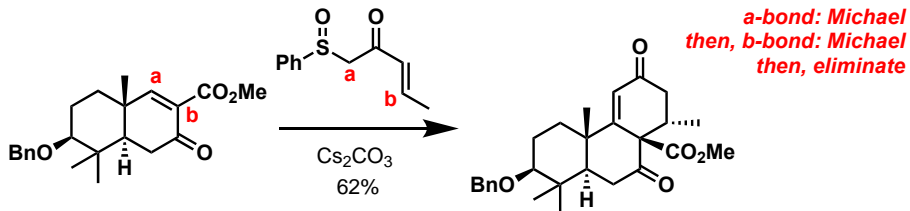


*c- and d-bonds:
Michael additions*

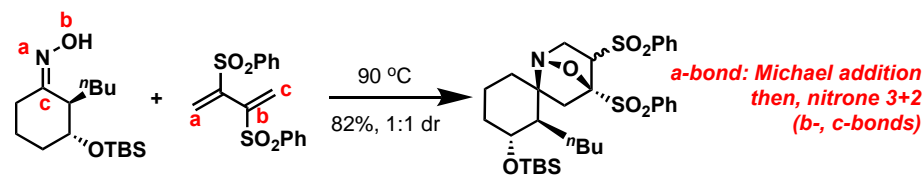
Rugulosin, Nicolaou et al., *Angew. Chem.* **2005**, 117, 8131.

Bin 1: Anionic cascades (cont.)

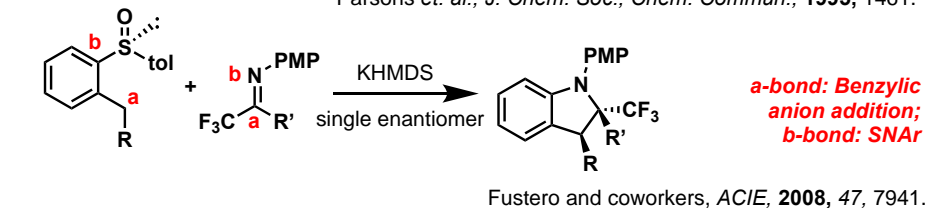
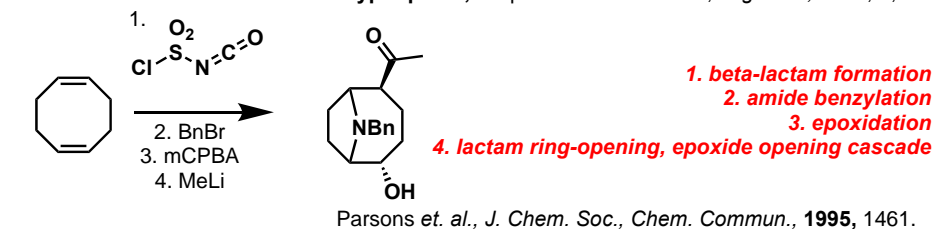
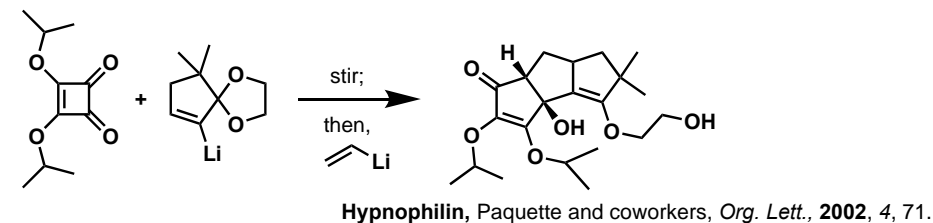
Michael / Oxa-Michael / Aldol reactivity (cont.)



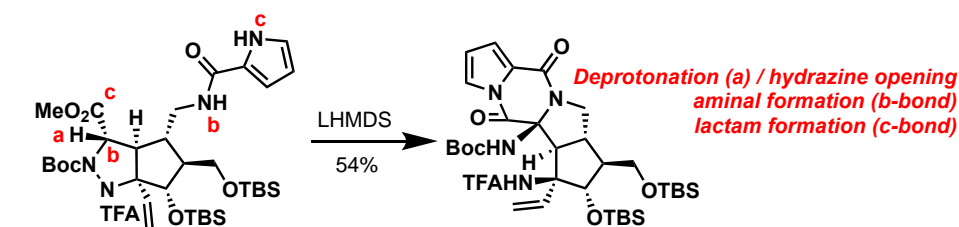
Michael / Oxa-Michael / Aldol reactivity (cont.)



Anionic cascades initiated by carbonyl additions

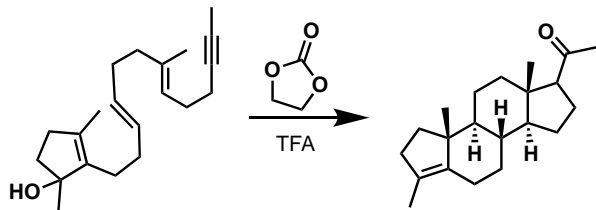


Nitrogen anion cascades

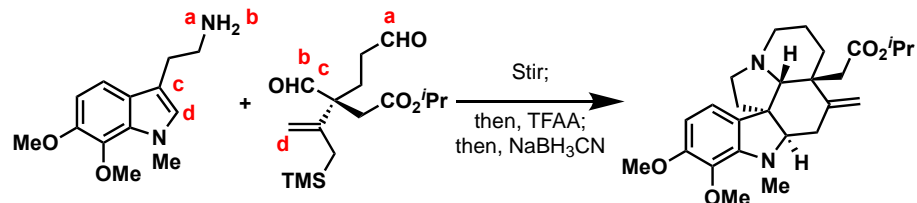


Bin 2: Cationic cascades

Polyene / olefin-cation-pi cascades

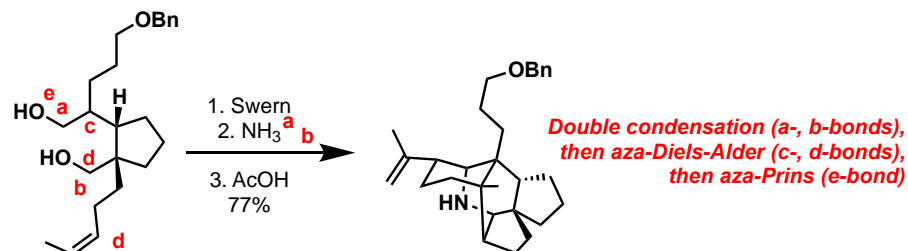


Covered extensively in other Baran group meetings;
See Polyene Cyclizations (Harwood, 2018)

Cationic cascades driven by reactive nitrogen electrophiles

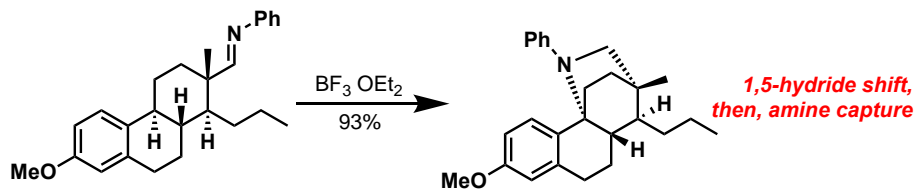
Double condensation (a-, b-bonds), then aza-Mannich (c-bond), then aza-Sakurai (d-bond), then imine reduction

Aspidophytine, Corey and coworkers, *J. Am. Chem. Soc.*, **1999**, 121, 6771.



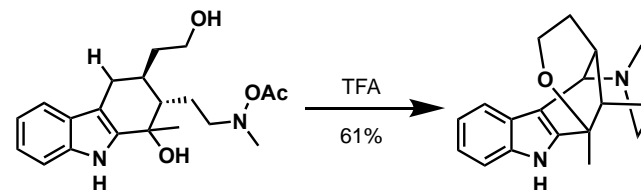
Double condensation (a-, b-bonds), then aza-Diels-Alder (c-, d-bonds), then aza-Prins (e-bond)

Secodaphniphylline, Heathcock *et al.*, *J. Org. Chem.*, **1992**, 57, 2566.

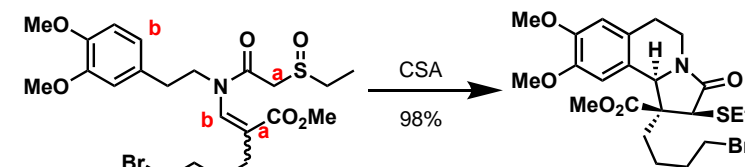


1,5-hydride shift, then, amine capture

Tietze and coworkers, *Angew. Chem. Int. Ed.*, **1999**, 38, 200.

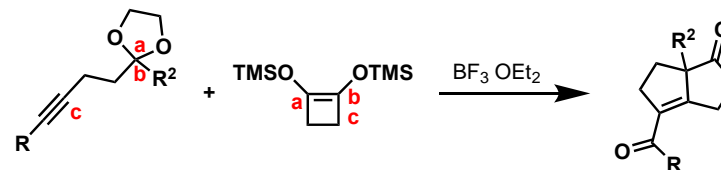
Cationic cascades driven by reactive nitrogen electrophiles (cont.)

Gilbertine, Blechert and coworkers, *J. Am. Chem. Soc.*, **2004**, 126, 3534.



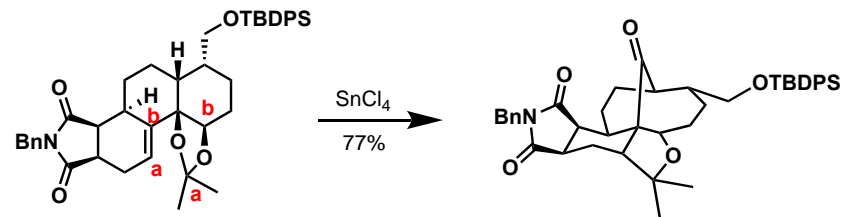
a-bond: enamine-Pummerer, then, b-bond: Pictet-Spengler

Jamtine, Padwa *et al.*, *Org. Lett.*, **2002**, 4, 715.

Pinacol-type cascades

Aldol (a-bond), then pinacol rearrangement (b-bond), then 5-exo-dig (c bond), then trapping by water

Curran and coworkers, *J. Org. Chem.*, **1995**, 60, 337.

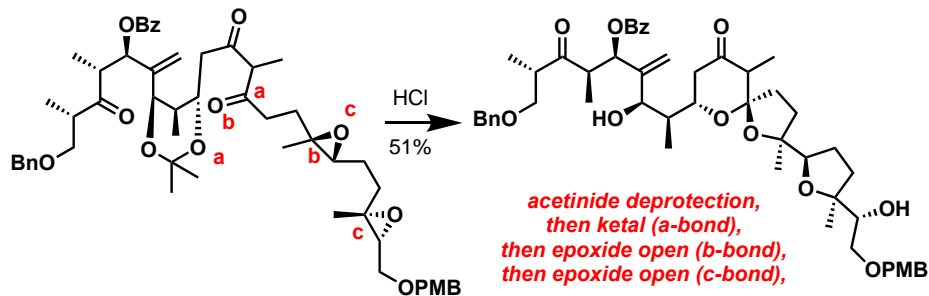
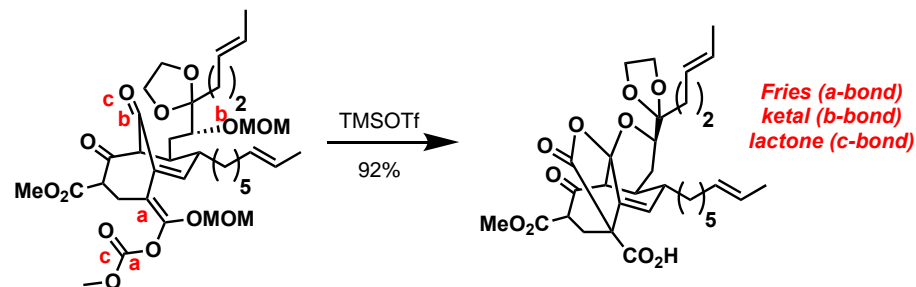
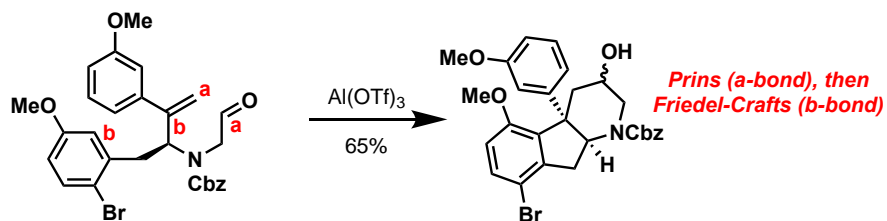
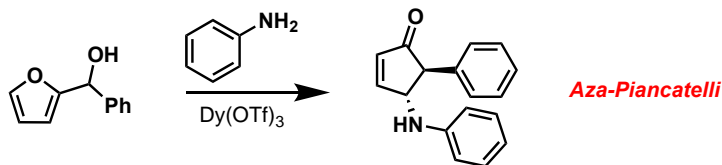
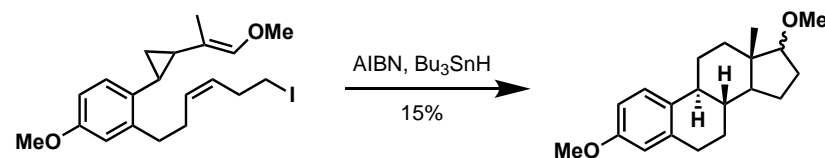
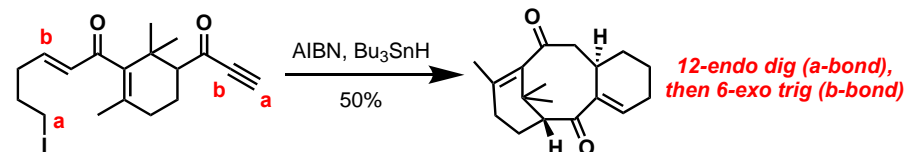
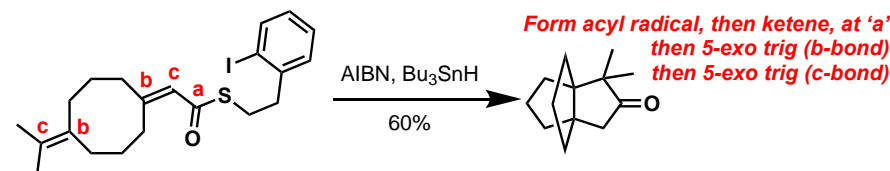
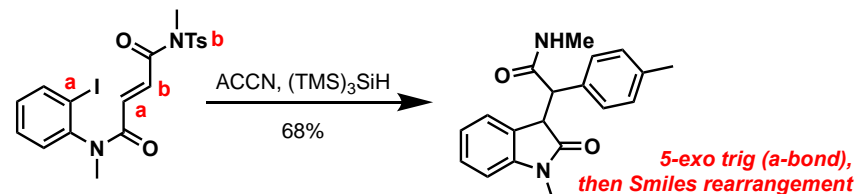


Prins (a-bond), then, pinacol rearrangement (b-bond)

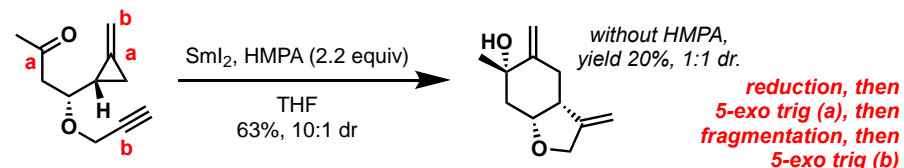
Barriault and coworkers., *Org. Lett.*, **2005**, 7, 5921.

Bin 2: Cationic cascades (cont.)

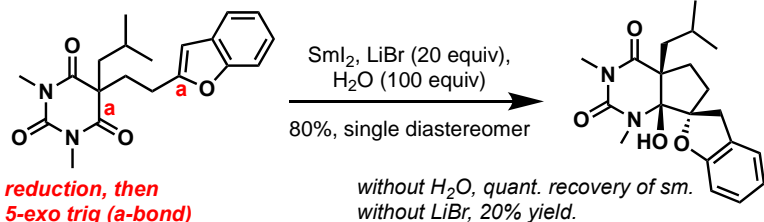
Cationic cascades driven by protonated oxygen electrophiles

Etheromycin, Paterson et al., *Tetrahedron Lett.*, **1993**, 34, 7137.CP molecules, Shair and coworkers, *J. Am. Chem. Soc.*, **2000**, 122, 7424.Hahuamine A, Aubé and coworkers, *Org. Lett.*, **2011**, 13, 2614.Read de Alaniz and coworkers, *Angew. Chem. Int. Ed.*, **2010**, 49, 9484.**Bin 3: Radical cascades**Radical chains (AIBN / R₃SnH or equivalent)For a classic, see Hirsutene, Curran and coworkers, *J. Am. Chem. Soc.* **1985**, 107, 1448.Oestrone, Pattenden et al., *Tetrahedron Lett.*, **2004**, 45, 4027.Towards Taxanes, Pattenden and coworkers, *J. Chem. Soc., Perkin. Trans. 1*, **1998**, 3181.Pattenden and coworkers, *Org. Biomol. Chem.*, **2005**, 3, 340.Sapi and coworkers, *Chem. Commun.* **2012**, 48, 2442.

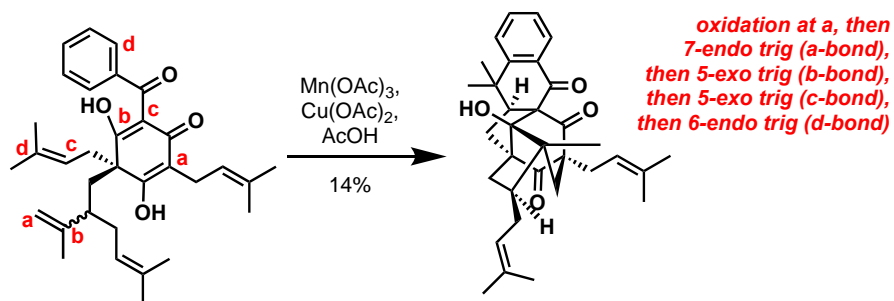
Radical cascades driven by reductions

Killburn and coworkers, *Chem. Commun.*, **1998**, 1875.

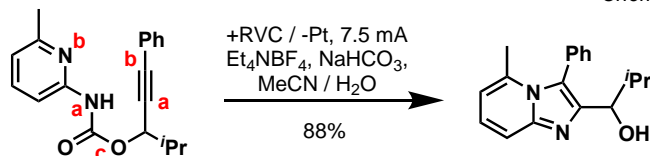
For more coverage of Sm-mediated cascades, see Baran group meeting, "Samarium (II) Iodide" (Lewis, 2010)

Bin 3: Radical cascades (cont.)**Radical cascades driven by reductions (cont.)**Procter and coworkers, *J. Am. Chem. Soc.*, **2017**, 139, 1661.

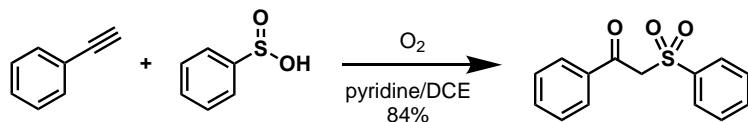
Ti(III) also common initiating reductant. See Baran group meeting "Organotitanium Chemistry" (Merchant, 2017)

Radical cascades driven by oxidationsGarcibracteatone, George and coworkers, *Org. Lett.*, **2012**, 14, 5162.

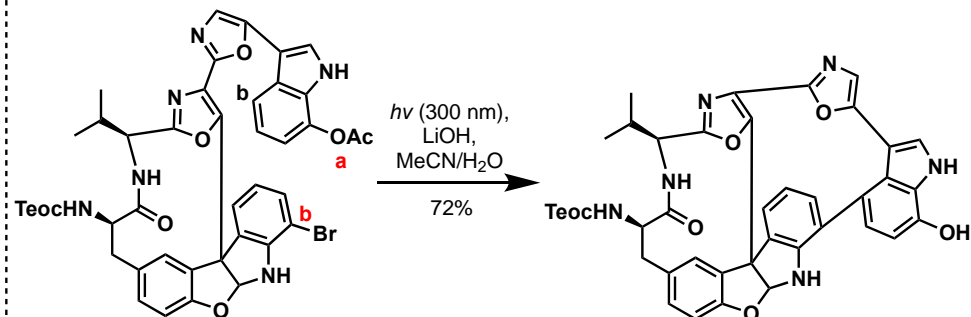
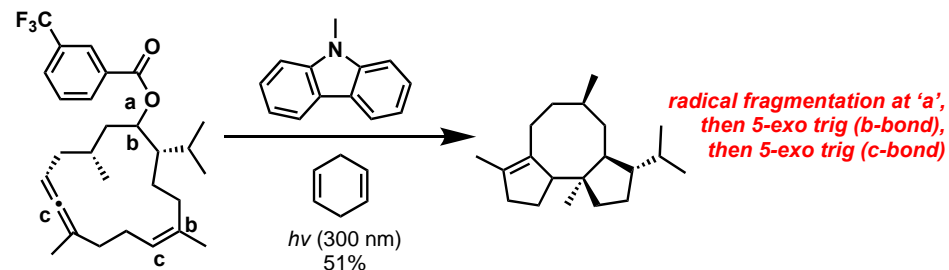
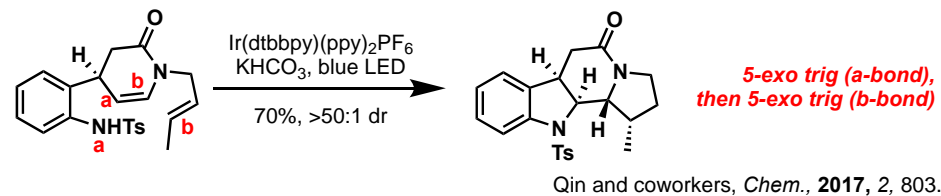
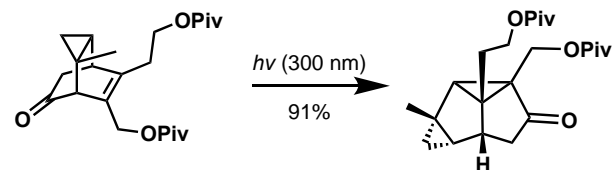
Extensive coverage of this pathway in Baran group meeting "Manganese (III) in Organic Chemistry" (Voica, 2010)



5-exo dig (a-bond), then 5-endo trig (b-bond), then radical cleavage of c-bond

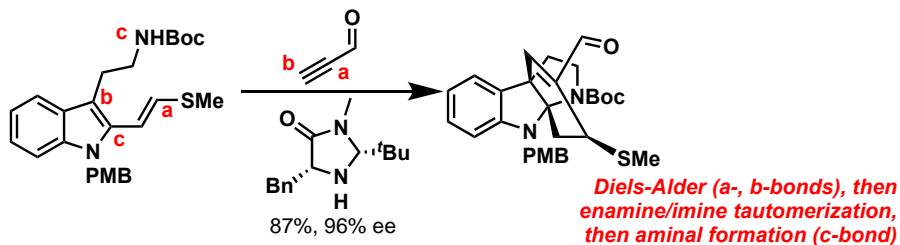
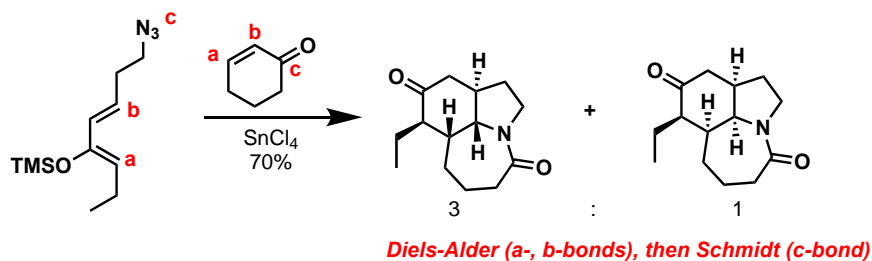
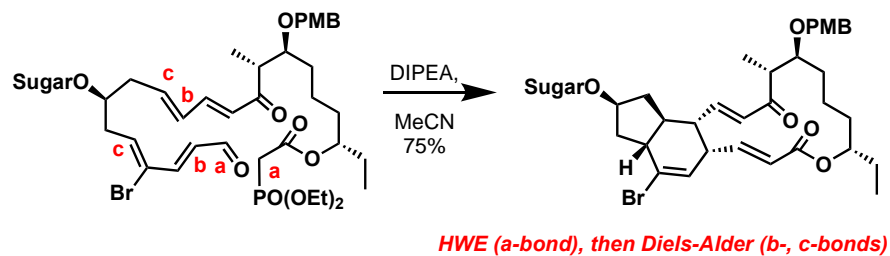
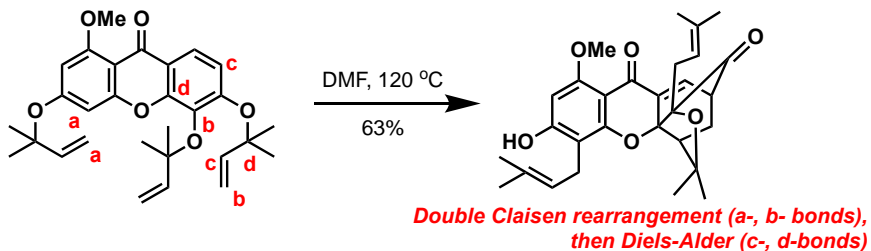
Xu and coworkers, *Angew. Chem.*, **2018**, 130, 1652.

oxidation of sulfinate, then addition into alkyne, then trapping with dioxygen, then reduction of peroxide with sulfinate

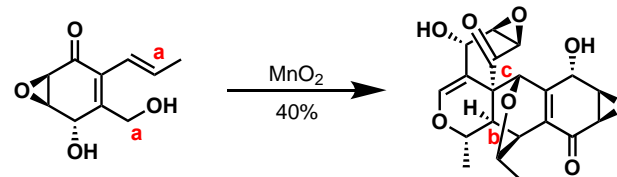
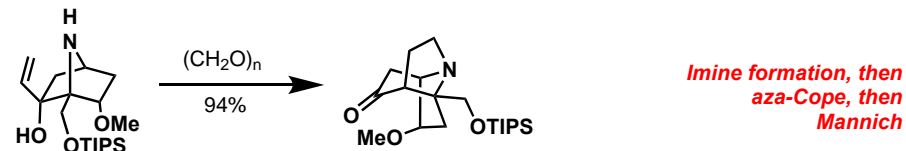
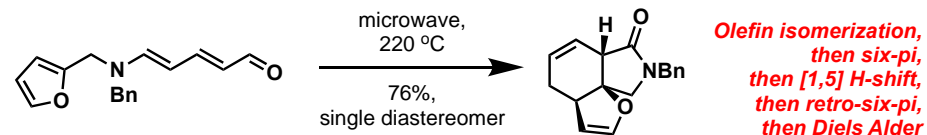
Lei and coworkers, *J. Am. Chem. Soc.* **2013**, 135, 11481.**Radical cascades driven by photoexcitation**hydrolysis of phenol (a), then biradical formation,
then combination (b-bond), then elimination of bromideDiazonamide A, Harran and coworkers, *Angew. Chem.*, **2003**, 115, 5111.Myers and coworkers, *J. Am. Chem. Soc.*, **1993**, 115, 7926.Qin and coworkers, *Chem.*, **2017**, 2, 803.Paquette and coworkers, *J. Am. Chem. Soc.*, **1991**, 113, 9384.

Bin 4: Pericyclic cascades

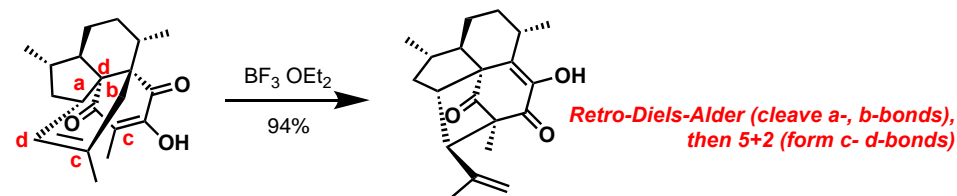
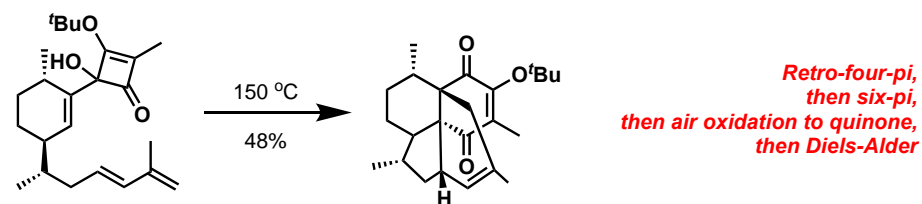
Pericyclic cascades driven by exergonic cycloadditions

Minfiensine, MacMillan and coworkers, *J. Am. Chem. Soc.*, **2009**, 131, 13606.Stenine, Aubé and coworkers, *J. Am. Chem. Soc.*, **2005**, 127, 15712.Spinosyn, Roush and coworkers, *Proc. Natl. Acad. Sci.*, **2004**, 101, 11955.1-O-methylforbesione, Nicolaou et al., *Angew. Chem. Int. Ed.*, **2001**, 40, 4264.

Pericyclic rearrangement cascades driven by net bond formation

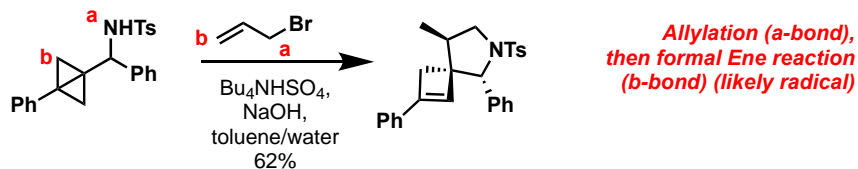
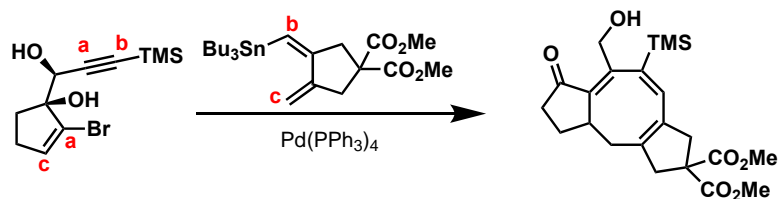
**Oxidation, then 6-pi electrocyclization (a-bond), then Diels-Alder dimerization (b-, c-bonds)**
Epoxyquinol A, Hayashi and coworkers, *Angew. Chem. Int. Ed.* **2002**, 41, 3192.Asparagine A, Overman and coworkers, *J. Am. Chem. Soc.*, **2003**, 125, 15284.Vanderwal and coworkers, *J. Am. Chem. Soc.*, **2009**, 131, 7546.

Pericyclic cascades driven by strain release

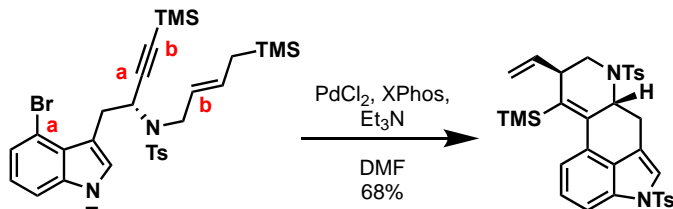
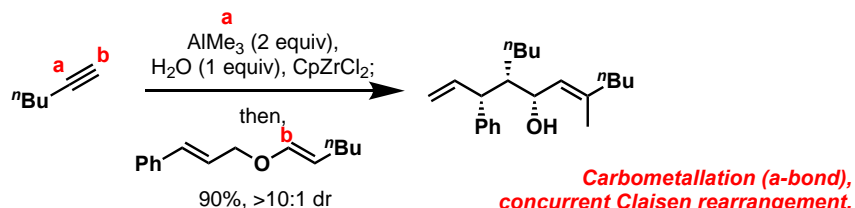
Elisapterosin B, Jacobsen and coworkers, *Angew. Chem. Int. Ed.*, **2005**, 44, 6046.Elisapterosin B, Harrowven et al., *Angew. Chem. Int. Ed.* **2005**, 44, 1221.

Bin 4: Pericyclic cascades (cont.)

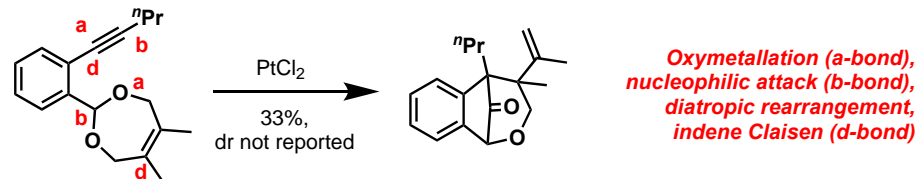
Pericyclic cascades driven by strain release (cont.)

Wipf et al., *Angew. Chem. Int. Ed.*, **2006**, 45, 4172.Towards Ophiobolin A, Suffert and coworkers, *Angew. Chem. Int. Ed.* **2004**, 43, 2826.**Bin 5: Transition metal-catalyzed cascades**

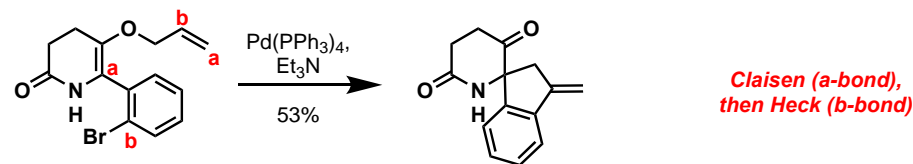
Carbometallation cascades

Lysergol, Werz and coworkers, *Org. Lett.*, **2017**, 19, 1914.Wipf et al., *J. Org. Chem.*, **2005**, 70, 8096.**Bin 5: Transition metal-catalyzed cascades (cont.)**

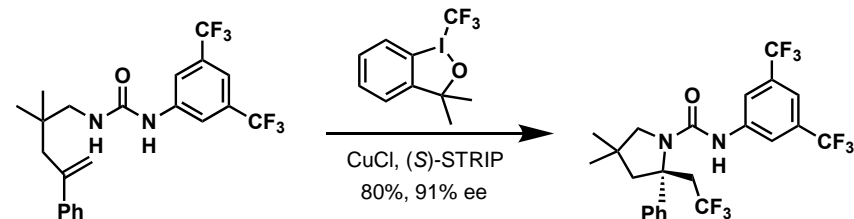
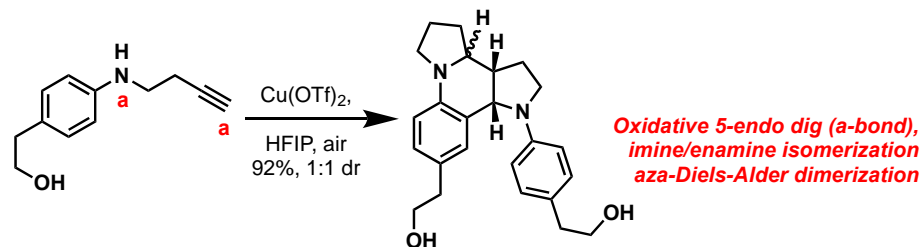
Nucleometallation cascades

Yamamoto and coworkers, *Chem. Lett.* **2005**, 34, 174.

Isomerization cascades

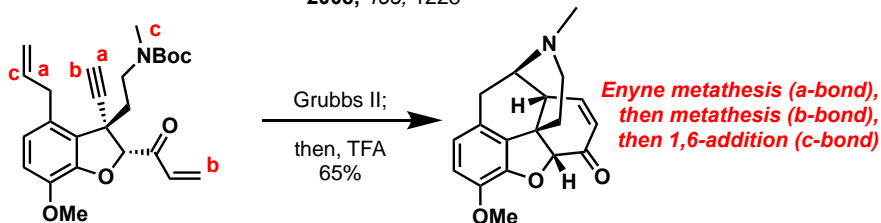
Watson et al., *Tetrahedron Lett.*, **1994**, 35, 9763.

Metal-catalyzed radical/SET cascades

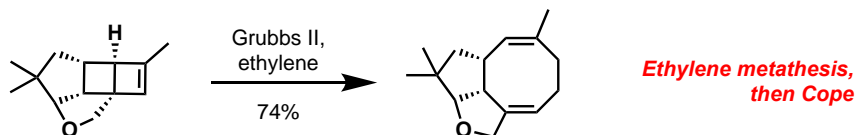
Liu and coworkers, *J. Am. Chem. Soc.*, **2016**, 138, 9357.Li and coworkers, *Eur. J. Org. Chem.*, **2016**, 3684.

Bin 5: Transition metal-catalyzed cascades (cont.)**Metathesis cascades**

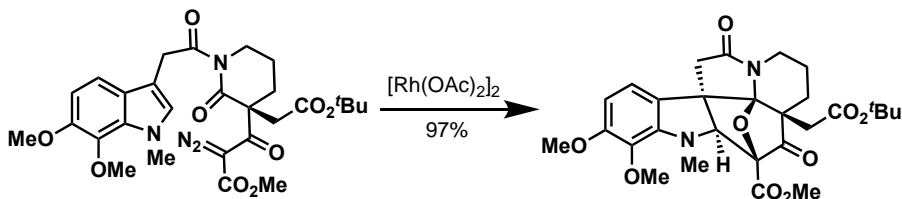
For a classic example, see *Cyanthiwign F synthesis*; Stoltz and coworkers, *Nature*, **2008**, 453, 1228



Morphine, Smith and coworkers, *Angew. Chem. Int. Ed.*, **2016**, 55, 14306.

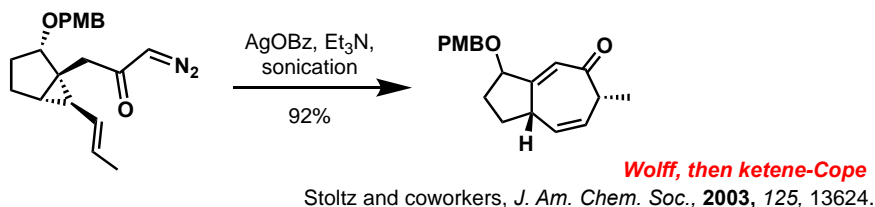


Asteriscanolide, Snapper and coworkers, *J. Am. Chem. Soc.*, **2000**, 122, 8071.

Carbenoid cascades

Carbonyl ylide formation, then 3+2

Aspidophytine, Padwa and coworkers, *Org. Lett.*, **2006**, 8, 3275.



Stoltz and coworkers, *J. Am. Chem. Soc.*, **2003**, 125, 13624.

Cascade reaction utility: A cheminformatics approach****Cascade efficiency as measured by yields**

Average yield of a Michael reaction (random selection, $n = 250$): **84%**

Average yield of a Michael-Michael cascade ($n = 26$): **56%**
(75% "per step")

Cascade efficiency as measured by usage in process chemistry

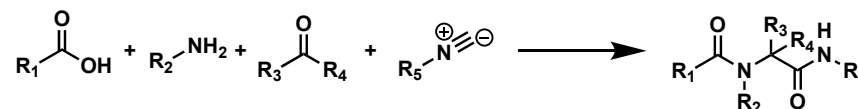
Percentage of 2017 OPRD syntheses that use cascade reactions: **7%**

Compare to Pd-catalyzed cross couplings: **81%**

Compare to amine acylations: **89%**

What were the cascades that were used in industry?

Almost exclusively Ugi reactions!



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**The following statistics were obtained using alpha-version cheminformatics data-mining software. Results are unvalidated and may be highly inaccurate.